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ANSWER/COMPREHENSION

QUESTION- PART ONE

Inside the Cell Membrane *How to Answer a 4-Mark*

Comparison Question:

Comprehension (Section I)

What is Inquiry-Based

Learning? *Protein Synthesis*

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Classroom Orbitals: Crash

Course Chemistry #25

Biology: Cell Structure I

Nucleus Medical Media **Why I**

Flipped My Classroom DNA vs

RNA (Updated)

The Periodic Table: Atomic

Radius, Ionization Energy,

and Electronegativity Taking

POGIL Activities On-Line in

Middle School Mitosis vs.

Meiosis: Side by Side

Comparison ~~Cell Transport~~

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Atomic Hook-Ups - Types of
Chemical Bonds: Crash Course
Chemistry #22 Photosynthesis
and Cellular Respiration
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Pogil 1. Name 3 strong acids
and write their formulas.
What makes an acid strong?

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Weak Acids
+hydrochloric acid HCl ,
nitric acid HNO_3 , sulfuric
acid H_2SO_4 . Strong means
100% ionized, all the H ions
have broken off the acid
molecule in water. It exists
as all ions. 2. Name 4 weak
acids and write their
formulas.

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2 POGIL™ Activities for High School Biology 5. All the internal structures are suspended (floating) in what substance? 6. One of the bacteria in Model 1 has a tail-like structure. a. What is this structure called? b. What might be the purpose of this structure? c. Based on your answer to the previous question, what might you infer about the ...

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What With strong acids this
is easy Hydrochloric acid is
a strong acid - virtually
100% ionised Each mole of
HCl reacts with the water to

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Weak Acids
give 1 mole of hydrogen ions
and 1 mole of chloride ions
That means that if the
concentration of

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Key. HNO_3 – Nitric Acid H_2SO_4 – Sulfuric Acid HClO_4 – Perchloric Acid Weak acids
do not dissociate
completely. Two common (AP
questions) weak acids are:
 HF – Hydrofluoric Acid HF
(aq) + H_2O (l) \leftrightarrow H_3O^+ (aq) + F^- (aq)
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