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CHAPTER 4 TEST: Atoms, Atomic

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Theory and Atomic Structure Matching.

A. Bohr B. Democritus C. Rutherford D.

Dalton E. Thomson F. Schrodinger _____

1. Greek thinker; called nature's basic particle an atom, based on the Greek word "atomos" which means "indivisible". Did not have evidence that atoms existed.

_____2.

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CHAPTER 4 TEST: Atoms, Atomic
Theory and Atomic Structure

4 Atomic Structure □ Chapter Test A.

Matching Match each description in
Column B with the correct term in Column
A. Column A Column B 1. proton 4.2 a. the
smallest particle of an element that retains

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the properties of that element 2. atom 4. }
b. the weighted average of the masses
of the isotopes of 3. mass number 4.5 an
element 4. atomic mass unit 4.5

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Structure - nlsd.k12.oh.us Chapter 4
(Atomic Structure) Test Study Guide An
atom is the smallest particle of an element
that retains the properties of that element.
Democritus lived from 460 B.C. \approx 370
B.C. and was among the first to suggest
the idea of atoms.

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Chapter 4 Atomic Structure Test B Answers

Find the volume, in liters, of a rectangular box 25 cm long, 10 cm wide, and 8 cm deep. (Chapter 3 Scientific Measurement.

Name Date ATOMIC STRUCTURE Class
Chapter Test B A. Matching Match each term in Column B with the dorrect

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description in Column A. Write the letter of the correct term on the line. Column B
electron mass number atomic number
atomic mass neutron atomic mass unit
proton isotopes atom periodic table N. X.
g. H. 1.

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Chemistry Atomic Structure Online Quiz
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has the same number of electrons as an
alpha particle? H. Li \square H \square He. Question 1
of 15. 2. According to Planck energy
travels in a discontinuous manner and it is
composed of large number of tiny discrete
units called: ...

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Chemistry Atomic Structure Online Quiz
Test MCQs

Questions : 1. The number of electrons in Na^+ (atomic number = 11) are (a) 11 (b) 10 (c) 12 (d) 13 2. There are four elements P, Q, R and S having atomic numbers of 4, 18, 10 and 16 respectively. The element

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which can exhibit covalency as well as electrovalency will be: (a)...

Chapter 4 Structure of Atom MCQ Test 2
Sciencel Class 9th ...

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Atomic Structure (Chapter Test A)

STUDY. Flashcards. Learn. Write. Spell.
Test. PLAY. Match. Gravity. Created by.
emily_reidt. Terms in this set (21) the total
number of protons and neutrons in the
nucleus of an atom. mass number. the

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weighted average mass of the atoms in a naturally occurring sample of an element.

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1 Sciencel Class 9th Questions: 1. The
four atomic species can be represented as
follows. Out of these, the two species

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which can be termed isobars are: (i)
201X60 (ii) 200X61 (iii) 200X58 (iv)
203X60 (a) (i) and (ii) (b) (ii) and (iii) (c)
(i) and (iii)...

Chapter 4 Structure of Atom MCQ Test 1
Sciencel Class 9th ...

4 Atomic Structure □ Chapter Test A.

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Matching Match each description in Column B with the correct term in Column A.

Column A	Column B
1. proton	4.2 a. the smallest particle of an element that retains the properties of that element
2. atom	4.3 b. the weighted average of the masses of the isotopes of an element
3. mass number	4.4 an element
4. atomic mass unit	4.5

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Chapter 4 Atomic Structure Test Answer
Key

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Chapter 4 Structure of ...

CHAPTER 4 TEST: Atoms, Atomic

Theory and Atomic Structure Chemistry

Chapter 4 Test - Atomic Structure. 1. The

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beam casted a shadow 2. The beam could spin a paddle wheel 3. The beam was deflected by a magnet 4. The beam was deflected by electrical fields 5. Different gases gave the same results - When
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Electronegativity.

MCQ on Structure of atom Class 11
Chemistry Chapter 2 ...

The chapter contains important topics such
as Atomic structure, atomic model,
Dalton's Atomic theory, Thomson Atomic

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Model, Rutherford Atomic theory, Bohr's Atomic theory, etc. The questions in this article help students to understand the difficulty level of questions in the upcoming JEE Main and JEE Advanced exams.

JEE Previous Year Question Bank on

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Chapter 4 (Atomic Structure) Test Study
Guide An atom is the smallest particle of
an element that retains the properties of
that element. Democritus lived from 460
B.C. – 370 B.C. and was among the first to

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suggest the idea of atoms.

Chapter 4 Atomic Structure Test Answer
Key

Read Free Chemistry Chapter 4 Atomic
Structure Test. Chemistry Chapter 4
Atomic Structure Transformed
Democritus's ideas of atoms into a

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scientific theory: 1) All elements are composed of tiny indivisible particles called atoms. 2) Atoms of the same element are identical. 3) Atoms of different elements can mix together forming mixtures of whole number ratio compounds. 4) Chemical reactions occur when atoms are separated, joined or

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rearranged.

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