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There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three states

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are solid, liquid, and gas. These three states are quite different. The main difference is in their particles.

Chapter 13: States of Matter - Chemistry by Anna

Section 13.4 Changes of State The relationship among the solid, liquid, and vapor states (or phases) of a substance in a sealed container are best represented in a single graph called a phase diagram Phase diagram- gives the temperature and pressure at which a substances exists as solid, liquid, or gas (vapor) 70.

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Chemistry Chapter 13 States Of Matter Chapter 13 States of Matter137 SECTION 13.1 THE NATURE OF GASES (pages 385–389) This section introduces the kinetic theory and

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describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386) 1.

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