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- Chapter 11 31 number of moles of known substance C H C H C H O O O

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Chemistry: Matter and Change •

Chapter 4 Math Skills Transparency Masters Calculating Atomic Mass MATH

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Use with Chapter 4, Section 4.3 4 Given the data in the table, calculate the atomic mass of unknown element X. Then, identify the unknown element, which is used

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**MATH SKILLS TRANSPARENCY MASTER.** Use with Chapter 4, Section 4.3. 4. Given the data in the table, calculate the atomic mass of unknown element X. Then, identify the unknown element, which is used medically to treat some mental disorders. Analyze the Problem. You are given the data in the table.

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Chapter 11, 1.2 **MATH SKILLS  
TRANSPARENCY WORKSHEET 1.**  
Explain what a crystalline solid is. 18 Use  
with Chapter 12, Section 2.3 1. The  
reaction of iron(III) oxide ( $\text{Fe}_2\text{O}_3$ ) and  
hydrogen ( $\text{H}_2$ ) is represented by the  
following unbalanced chemical equation.  
 $\text{Fe}_2\text{O}_3(s) + \text{H}_2(g) \rightarrow \text{Fe}(s) + \text{H}_2\text{O}(l)$   
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WORKSHEET Determining Numbers of  
Ions Use with Chapter 7, Section 7.3 1.  
Write a simple word equation that  
illustrates what must be true of total  
positive charge and total negative charge  
in an ionic compound. 2. How many  
potassium ions (group 1) would be related  
to balance the charge of each of the  
following in a compound?

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Snow Elementary School – Dearborn Public Schools

June 13th, 2018 - 94 Chemistry Matter and Change • Chapter 4 Math Skills Transparency Masters Calculating Atomic Mass MATH SKILLS

TRANSPARENCY MASTER Evaluate the Answer" MATH SKILLS

TRANSPARENCY MASTER 16

TRENHOLM SCIENCE MAY 21ST, 2018 - NAME DATE CLASS MATH SKILLS

Math Skills Transparency Answers

82 Chemistry: Matter and Change • Chapter 25 Math Skills Transparency Worksheets 1. In equation 1, what do the numbers 238 and 92, written to the left of the symbol U, represent? 2. Explain how both mass number and atomic number have been conserved in equation 1. 3. Write the nuclear symbol for the missing

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Chapter 13 1. 4.  
term in equation 2.

balancing nuclear equations

Math Skills Transparency Worksheets

Chemistry: Matter and Change •

Chapter 5 13 1. Look at the two waves

shown. What is the speed of each wave? 2.

Look at the two waves shown. Which wave has a higher frequency? Which wave has a longer wavelength? 3.

Assume that wave A has a wavelength of 699 nm. Calculate the frequency of the wave. Show your work. 4.

MATH SKILLS TRANSPARENCY

MASTER 5 Interpreting Waves Use ...

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**MATH SKILLS TRANSPARENCY WORKSHEET** Use with Chapter 25, Section 25.3 42 82 1. In equation 1, what do the numbers 238 and 92, written to the left of the symbol U, represent? 238 is the mass number of the isotope, and 92 is the atomic number of the isotope. 2. Explain how both mass number and atomic number have been conserved in equation 1.

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Worksheet Answers MATH SKILLS

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Determining Numbers of Ions Use with

Chapter 7, Section 73 1 Write a simple word equation that illustrates what must

be true of total positive charge and total negative charge in an ionic compound 2

How many potassium ions (group 1)

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