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I. Reactant Stability/Reactivity: The more stable the reactant, the less reactive it will be. In terms of rates, this means that the more stable the reactant, the slower it will react. (The concept

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of a ketone will produce. Reduction of a carboxylic acid will pro.... Reduction of a ketone will produce a (n) Carboxylic acid. No reaction. Aldehyde (or full reduction would be a primary alcohol. Secondary alcohol.

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IUPAC name for this structure?

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