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Chemistry Reactions and Reagents  
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chemistry mechanisms - revision  
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Sigma and Pi Bonds: Hybridization  
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groups~~ BCH1108 ORGANIC  
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1.1: The Origins of Organic Chemistry  
The term "organic chemistry" was first used in the 1800's to distinguish reactions performed by chemists from those occurring within living

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organisms. 1.2: Principles of Atomic Structure (Review) Atomic structure is applied to several topics of organic chemistry.

1: Introduction and Review -  
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Chapter 1 Introduction and Review 1

17) Provide a Lewis structure for a molecule with molecular formula  $\text{C}_2\text{H}_2\text{O}_2$ . Answer: Diff: 2 Section: 1.4 18)

Draw the Lewis structure for 2-butanol,  $\text{C}_4\text{H}_{10}\text{O}$ .

practicech1ans.doc - Organic

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Review 1) While you were up late one  
night studying organic chemistry, you  
happened to see the last 5 minutes of  
an infomercial on TV. The

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spokesperson claimed that their brand of automobile tires were superior to

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Jan William Simek - 6 ...

ORGANIC CHEMISTRY I CHEM 2323

Fall 2014 Instructor: Michael C.

Biewer biewerm@utdallas.edu, BE

3.326 Syllabus. Lecture Notes: The  
following material can be downloaded  
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Chapter 5 . Material for test #2 .  
Chapter 4

Organic Chemistry I  
Chapter 1 Wade Organic Chemistry.  
acid-dissociation constant ( $K_a$ )  $pK_a$ .  
Arrhenius acid. Arrhenius base. the

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equilibrium constant for the reaction  
of the acid with water... the negative  
logarithm of  $K_a$ . dissociates in water  
to give  $H^+$  and  $O^{2-}$ . dissociates in water to  
give  $OH^-$ .

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Chapter 2 Structure and Properties of  
Organic Molecules 1) An orbital can  
be described by its \_\_\_\_\_, which is the  
mathematical description of the shape  
of the electron wave as it oscillates.

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Answer: wave function Diff: 1 Section:  
2.1 2) The electron density at any  
point is proportional to the \_\_\_\_\_ of  
the electron ...

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12 - Infrared Spectroscopy and Mass

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Chemistry, 7th Edition L. G. Wade, Jr.  
Infrared Spectroscopy and Mass  
Spectrometry 2. Chapter 12 2  
Introduction • Spectroscopy is a  
technique used to determine the  
structure of a compound.



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12 - Infrared Spectroscopy and Mass Spectrometry - Wade 7th  
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1.1 The Origins of Organic Chemistry  
1.2 Principles of Atomic Structure 1.3  
Bond Formation: The Octet Rule 1.4  
Lewis Structures 1.5 Multiple Bonding

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1.6 Electronegativity and Bond  
Polarity 1.7 Formal Charges 1.8 Ionic  
Structures 1.9 Resonance 1.10  
Structural Formulas 1.11 Molecular  
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Chapter 14 Ethers, Epoxides, and Sulfides

1) What is the hybridization of the oxygen atom in dialkyl ethers?  
A)  $sp^3$  B)  $sp^2$  C)  $sp$  D)  $s$  E)  $p$  Answer: A Diff: 1 Section: 14.2

2) Which of the following corresponds to the C-O-C bond angle in dimethyl ether?  
A)  $60^\circ$  B)  $94^\circ$  C)  $110^\circ$  D)  $122^\circ$  E)  $180^\circ$

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Answer: C Diff: 1 ...

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Chapter 2 Acids and Bases; Functional  
Groups 1) An orbital can be described  
by its \_\_\_\_\_, which is the

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mathematical description of the shape of the electron wave as it oscillates.

Answer: wave function Diff: 1 Section: 1.12 LO: 2.1 2) The electron density at any point is proportional to the \_\_\_\_\_ of the electron ...

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